

Definition Unsaturated Solution Chemistry

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UNSATURATED SOLUTION DEFINITION - THOUGHTCO

Thu, 20 Apr 2017 23:58:00 GMT

unsaturated solution definition. an unsaturated solution is a chemical solution in which the solute concentration is lower than its equilibrium solubility.

UNSATURATED | DEFINE UNSATURATED AT DICTIONARY

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unsaturated definition, not saturated; having the power to dissolve still more of a substance. see more.

DEFINITION OF UNSATURATED SOLUTION IN CHEMISTRY

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(chemistry) (of a solution) containing less solute than a saturated solution. ?unsatu?ration n. un•sat•u•rat•ed ... "an unsaturated salt solution"

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... supersaturated solution, unsaturated. ... (chemistry) (of a solution) containing more solute than a saturated solution and therefore not in equilibrium. 2.

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in chemistry, saturation (from the latin word saturare, meaning 'to fill') has diverse meanings, all based on the idea of reaching a maximum capacity.

SATURATED SOLUTION DEFINITION AND EXAMPLES - THOUGHTCO

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this is the definition of saturated solution as the term is used in chemistry, as well as examples of saturated solutions.

SUPERSATURATED | DEFINE SUPERSATURATED AT DICTIONARY

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supersaturated definition, to increase the concentration of (a solution) beyond saturation; saturate abnormally. see more.

SATURATED SOLUTION: DEFINITION & EXAMPLES - STUDY

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definition of saturated solution. a saturated solution is one where there are about equal amounts of particles or solutes and solvent in the solution.

UNSATURATED SOLUTION - DEFINITION, FACTS & EXAMPLE ...

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we can observe many examples of un-saturated solutions in our everyday life. salt in water or sugar in water can be unsaturated solutions if they can dissolve more ...

WHAT IS THE DIFFERENCE BETWEEN A UNSATURATED, SATURATED ...

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what is the difference between a unsaturated, saturated, and supersaturated solutions? what is a solution? first of all you should know what a solution is.

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an unsaturated solution describes a solution that has a less than the max amount of solute ... what are saturated, unsaturated and supersaturated solutions? chemistry.

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an unsaturated solution is one in which more solute can be dissolved, i.e, it's saturation point has not been reached.

WHAT ARE SOME EXAMPLES OF SUPERSATURATED SOLUTIONS ...

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what are some examples of supersaturated solutions? chemistry solutions saturated and supersaturated ... what is an unsaturated hydrocarbon with a triple bond and two ...

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